Application Sheet io·li·tec



Ionic Liquids Technologies

Electrodeposition of Aluminum

Due to the limited electrochemical window of water, aqueous solutions cannot be used for the electrodepostion of less noble metals like aluminum. Nevertheless, there are many interesting applications for aluminum layers as corrosions inhibitors or design elements. Brilliant aluminum layers can i. e. replace chromium layers. Therefore, the electrodeposion of aluminum from waterfree electrolytes is of high technical interest.

Ionic liquids are low melting salts, which are liquid below 100 °C. They are incombustible and posssess a high thermal stability, a low vapour pressure as well as good ionic conductivities. Their wide electrochemical windows (up to 6 V) makes them suitable media for the electrodeposition of metals like aluminum. Therefore, ionic liquids based electrolytes are an attractive alternative to aqueous electrolytes. Since their price is naturally above water-based electrolytes, it's important to know that ionic liquids can be recycled easily and reused.

IOLITEC developed ready-to-use electrolytes for the alumium deposition from which you are able to deposite brilliant aluminum layers on copper (Fig. 1), steel (Fig. 2) or aluminum (Fig. 3). There are nearly no limitations on the structure of the substrate (Fig. 4-6).

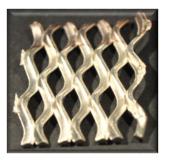


Fig. 4: Aluminium-coated steel wire mesh



Fig. 5: Aluminium-coated toy car

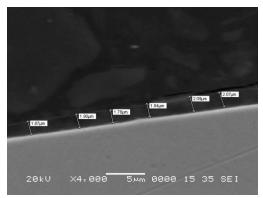


Fig. 6: SEM cross section.

The SEM micrograph aluminum the on the surface copper plate shows a dense aluminum which has layer, after a deposition

layer time of 10 min at room temperature a thickness of approximately 2 µm.

IOLITEC offers the following electrolytes for the aluminum deposition as catalogue products. We are also able to develop customized electrolytes.

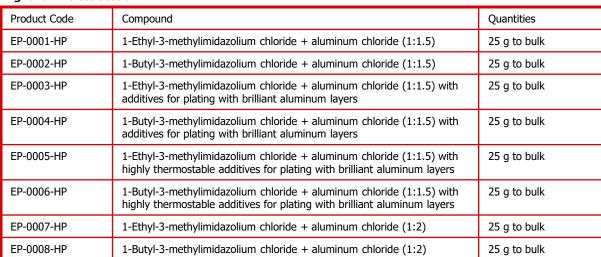




Fig. 1: Aluminium-coated copper plate



Fig. 2: Aluminium-coated steel plate



Fig. 3: Aluminium-coated alumium plate with zinc as intermediate





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